

## 2-METHOXY-1,2-DIHYDRORHAZIMINE, AN ALKALOID FROM LEAVES OF *RHAZYA STRICTA*

ATTA-UR-RAHMAN and SAJIDA KHANUM

H. E. J. Research Institute of Chemistry, University of Karachi, Karachi-32, Pakistan

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**Key Word Index**—*Rhazya stricta*; Apocynaceae; alkaloids; rhazimine; NMR

**Abstract**—Studies on the alkaloids from the leaves of *Rhazya stricta* have afforded a new alkaloid to which the structure 2-methoxy-1,2-dihydrorhazimine has been assigned.

### INTRODUCTION

*Rhazya stricta* belongs to the family Apocynaceae. It is abundantly distributed in various parts of Pakistan [1–3] and is widely used in the treatment of several diseases [4, 5]. It is specially reputed for its anti-tumour activities. A number of cytotoxic alkaloids have previously been reported from the plant [6, 7].

### RESULTS AND DISCUSSION

The crude alkaloidal material isolated by conventional procedures [8, 9] was subjected to chromatographic separation to afford a substance which possessed a typical indoline UV spectrum. The IR spectrum (chloroform) showed the presence of ester and ketonic carbonyl groups and an imine group. High resolution mass spectrometry afforded  $[M]^+$  at  $m/z$  382.1895 which agreed with the mass calculated for the formula  $C_{22}H_{26}N_2O_4$  (382.1893) indicating the presence of 11 double bond equivalents in the molecule. The base peak at  $m/z$  122.0965 corresponded to the formula  $C_8H_{12}N$  attributed to the ion (d) [10] which could arise by cleavage between the C-5 and C-16 bonds. The facile loss of methanol from the  $[M]^+$  suggested the presence of a methoxy group. The  $^1H$  NMR spectrum (deuteriochloroform) showed a three-proton doublet at  $\delta$  1.52 ( $J_1 = 7$  Hz,  $J_2 = 2.5$  Hz) which was assigned to the ethylidene methyl group. A three-proton singlet at  $\delta$  3.14 was assigned to the methyl protons of the methoxy group while another three-proton singlet at  $\delta$  3.53 was assigned to the ester methyl group. A downfield one-proton singlet at  $\delta$  4.93 was attributed to the C-2 proton. The olefinic proton of the ethylidene group resonated as a quartet at  $\delta$  5.5 ( $J = 7$  Hz). The aromatic protons appeared as complex multiplets in the region  $\delta$  6.5–7.5.

The substance was found to be highly labile, being readily transformed to a faster moving substance when kept in chloroform at 30° for 2–3 hr. The  $^1H$  NMR spectrum of the transformation product showed the disappearance of the one-proton singlet for the C-2 proton at  $\delta$  4.93 and the appearance of a low field singlet at  $\delta$  7.70 due to the olefinic proton of the ketimine system in rhazimine [10]. This transformation product was identified as rhazimine [10], previously reported by us from the same plant, by direct comparison with an authentic

sample (co-chromatography, mp, IR, UV,  $^1H$  NMR, mass spectrum). The ready transformation of the indoline to rhazimine bearing an indolenine chromophore strongly supported the conclusion that the methoxy group was located at C-2. The lack of a bond between C-2 and C-3 in rhazimine had previously been confirmed by gated spin-echo measurements which had established the presence of a proton on C-2 (ketimine), and showed that C-3 and C-21 were both  $CH_2$  groups [10–12]. The stereochemistry of **1** is not known.

The  $^{13}C$  NMR (deuteriochloroform) of the alkaloid and its comparison with rhazimine is shown in Table 1. On the basis of these data, structure **1** is assigned to the alkaloid. The presence of **1** in the crude plant extract before contact with methanol was ascertained by TLC. This showed that **1** is a genuine natural product and not an artefact of isolation. The substance probably arises in the plant by hydration and subsequent methylation of rhazimine.

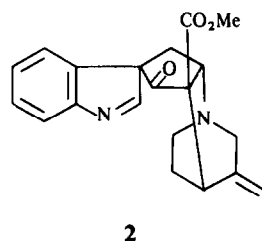
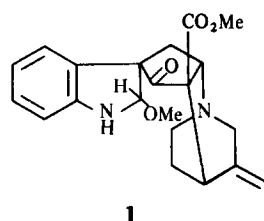


Table 1.  $^{13}\text{C}$ -NMR spectral data of 2-methoxy-1,2-dihydrorhazimine (1) and rhazimine (2)

Carbon No.	2-Methoxy-1,2-dihydrorhazimine	Rhazimine
2	85.93	60.55
3	48.41	51.55
5	57.02	61.01
6	27.30	30.25
7	53.02	63.07
8	127.10	137.46
9	128.07	128.23
10	119.75	125.00
11	115.34	127.88
12	127.75	128.69
13	143.01	142.70
14	22.25	31.93
15	36.93	37.24
16	54.02	58.07
17	*	214.11
18	12.75	12.86
19	119.38	120.84
20	130.89	137.32
21	51.74	52.92
$\text{OCH}_3$	50.74	—
Ester $\text{C}=\text{O}$	†	168.33
Ester $\text{OMe}$	51.81	52.01

\*†Signals too weak to be detected. In the transformation product, rhazimine, the corresponding signals 'a' and 'b' appeared at  $\delta$ 214.71 (ketone) and 168.33 (ester carbonyl), respectively

#### EXPERIMENTAL

**Isolation of 2-methoxy-1,2-dihydrorhazimine.** The crude alkaloidal material (170 g) isolated by the previously reported procedure [8, 9] from leaves of *R. stricta* Decaisne (45 kg) was subjected to flash chromatography over silica gel. The fraction obtained on elution with petrol- $\text{CHCl}_3$  (4:5) was concd to a gum (18 g) and again subjected to flash chromatography over silica gel. The fraction obtained on elution with  $\text{CHCl}_3$ -MeOH (25:3) afforded a mixture of five alkaloids, which was again loaded onto another silica gel column. Elution with mixtures of  $\text{CHCl}_3$ -MeOH of increasing polarity afforded a number of fractions. The fraction obtained on elution with  $\text{CHCl}_3$ -MeOH (10:1) afforded an alkaloid which was purified by prep. TLC on

$\text{Al}_2\text{O}_3$  (Merck, Type E) to afford 52 mg of a white crystalline (hygroscopic) material which gave a dark pink colouration with  $\text{CeSO}_4$ ,  $[\alpha]_D^{25}$  ( $\text{CHCl}_3$ ) +85°. IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3400 (NH), 1745 (keto  $\text{C}=\text{O}$ ) and 1720 (ester  $\text{C}=\text{O}$ ); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 210, 249 and 295;  $\lambda_{\text{min}}^{\text{MeOH}}$  nm: 232 and 275;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$ 1.52 (3H, dd,  $J_1 = 7$  Hz,  $J_2 = 2.5$  Hz,  $\text{CH}_3$ -HC=C), 3.14 (3H, s,  $-\text{OCH}_3$ ), 3.53 (3H, s, C- $\text{OCH}_3$ ), 4.97 (1H, s, H-2), 5.5 (1H, q,  $J = 7$  Hz  $>\text{C}=\text{CH}-\text{CH}_3$ ), 5.5-7.5 (4H, m, ArH); high resolution MS: 382.1895 ( $[\text{M}]^+$ , 48% calc. for  $\text{C}_{22}\text{H}_{26}\text{N}_2\text{O}_4$ : 382.1893); 367.1672 (9.8% calc. for  $\text{C}_{21}\text{H}_{23}\text{N}_2\text{O}_4$ : 367.1658); 350.1629 (15% calc. for  $\text{C}_{21}\text{H}_{22}\text{N}_2\text{O}_3$ : 350.1630  $[\text{M}-\text{MeOH}]^+$ ); 323.1759 (15% calc. for  $\text{C}_{20}\text{H}_{23}\text{N}_2\text{O}_2$ : 323.1759); 263.1546 (16% calc. for  $\text{C}_{18}\text{H}_{19}\text{N}_2$ : 163.1548); 218.1175 (48% calc. for  $\text{C}_{13}\text{H}_{16}\text{NO}_2$ : 218.1181); 182.0603 (24% calc. for  $\text{C}_{12}\text{H}_8\text{NO}$ : 182.0606) [13]; 167.0693 (18% calc. for  $\text{C}_{12}\text{H}_8\text{N}$ : 167.0734) [14]; 122.0966 (100% calc. for  $\text{C}_8\text{H}_{12}\text{N}$ : 122.0969).

**Conversion of 2-methoxy-1,2-dihydro-rhazimine (1) to rhazimine (2).** 2-Methoxy-1,2-dihydrorhazimine (13 mg), was dissolved in  $\text{CHCl}_3$  and allowed to stand for 2 hr at 30°. TLC on silica gel  $\text{CHCl}_3$ -MeOH (17:3) showed the formation of a faster moving spot which was separated by prep. TLC and crystallized from  $\text{CHCl}_3$ -MeOH (4:1) as colourless needles. The product was identified as rhazimine by direct chromatographic and spectral comparison (IR, UV,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, MS) with an authentic sample.

#### REFERENCES

- Hooker, J. D. and Jackson, B. D. (1865) *Indexkewensis*, Vol. 4, p. 705 and suppl. 8 (1926-30). Clarendon Press, Oxford.
- Bisset, N. G. (1958) *Ann. Bogor*, 3, 170.
- Hooker, J. D. (1875) *Flora of British India*, Vol. 3, p. 540. Reeve.
- Anon (1926) *Curtis Bot. Mag.* 152, 9119.
- Atta-ur-Rahman and Fatima, K. (1983) *Phytochemistry* 22, 1017.
- Mukhopadhyay, S., Handy, G. A., Funayama, S. and Cordell, G. A. (1981) *J. Nat. Prod.* 44, 896.
- Siddiqui, S. and Bukhari, A. Q. S. (1972) *Nature* 235, 393.
- Ahmad, Y., Fatima, K., Atta-ur-Rahman, Occolowitz, J. L., Solheim, B. A., Clardy, J., Garnick, R. L. and LeQuesne, P. W. (1977) *J. Am. Chem. Soc.* 99, 1943.
- Atta-ur-Rahman and Khanum, S. (1984) *Phytochemistry* 23, 709.
- Atta-ur-Rahman and Khanum, S. (1984) *Tetrahedron Letters* 25, 3913.
- (Varian) Bhacca, N. S., Hollis, D. P., Johnson, L. F. and Pier, E. A. (1962-1963) *High Resolution NMR Spectra Catalogue*.
- Shamma, M. and Hidenlang, D. M. (1979) *C-13 NMR Shift Assignments of Alkaloids*. Plenum Press, New York.
- Hesse, M. (1974) *Indolalkaloide*, Teil 1: Text p. 178. Verlag Chemie, Weinheim.
- Biemann, K., Bommer, P., Burlingame, A. L. and McMurray, W. J. (1964) *J. Am. Chem. Soc.* 86, 4624.